## An Unexpected Rearrangement of 4-Alkylaminoindoles

## Steven V. Ley\* and Roderick A. Porter

Department of Chemistry, Imperial College, London SW7 2AY, U.K.

4-Alkylaminoindoles rearrange in good yield to the corresponding 1-alkyl-4-aminoindoles in the presence of 10 mol% hydrated toluene-*p*-sulphonic acid in boiling toluene.

During studies directed towards the total synthesis of teleocidin<sup>1</sup>A (1) (lygnbyatoxin<sup>2</sup>) and related potent tumour promotors,<sup>3</sup> we observed a novel and unexpected rearrangement of 4-alkylaminoindoles, the details of which we report here.

The starting 4-alkylaminoindoles (2) used in this work were prepared from 4-aminoindole which in turn can be prepared by the excellent Leimgruber and Batcho method.<sup>4</sup> Compound (2, R = Me) was obtained in 78 % yield by lithium aluminium hydride reduction of 4-N-formylaminoindole while (2, R =  $CH_2Ph$ ) and (2, R =  $CH_2CO_2Et$ )† were derived by monoalklyation of 4-aminoindole using potassium carbonate-potassium iodide and the appropriate bromide, in 47 and 84% yields respectively.

† All new compounds were fully characterised by spectroscopic methods, and accurate mass and/or microanalytical techniques.

Upon heating the indoles (2) in toluene containing 10 mol% of monohydrated toluene-*p*-sulphonic acid smooth conversion into the corresponding l-alkyl-4-aminoindoles (3)† was achieved (Table 1). In the absence of water, or using anhydrous camphorsulphonic acid in a similar manner, no rearrangement was observed even after extended periods of time.

We suggest that the mechanism for the rearrangement reaction therefore involves initial ring opening of the 4-alkylaminoindole to produce an intermediate species (4) which prefers to undergo ring closure to the more thermodynamically stable 1-alkyl-4-aminoindole system (Scheme 1). This re-

Scheme 1

Table	1
-------	---

	Product (3)	Reaction time
Starting indole (2)	% yield	/h
R = Me	75	23
$R = CH_2Ph$	90	21
$R = CH_2CO_2Et$	77ª	44

<sup>a</sup> Plus 13% recovered starting material.

arrangement has potentially useful synthetic applications for the preparation of specifically substituted indoles.

We thank the S.E.R.C. and Pfizer Central Research, Sandwich, Kent, for a C.A.S.E. research studentship (to R. A. P.), the Royal Society of Chemistry for the Hickin-

bottom Research Award (to S. V. L.), and Dr. J. C. Ruddock (Pfizer, U.K.) for helpful discussions.

Received, 21st September 1982; Com. 1123

## References

- M. Takashima and H. Sakai, Bull. Agric. Chem. Soc. Jpn., 1960, 24, 647, 652; M. Takashima, H. Sakai, and K. Arima, Agric. Biol. Chem., 1962, 26, 660.
- 2 J. H. Cardellina II, F-J. Marner, and R. E. Moore, Science, 1979, 204, 193.
- 3 T. Sugimura, F. Hirota, M. Mori, M. Nakayasu, M. Terada, K. Umezawa, and R. E. Moore, *Carcinog. Comp. Serv.*, 1982, 7, 69 (*Chem. Abs.*, 96, 211 864).
- 7, 69 (Chem. Abs., 96, 211 864).
  4 A. D. Batcho and W. Leimgruber, U.S. Patent 3,967,639 (1976);
  L. I. Kruse, Heterocycles, 1981, 16, 1119.